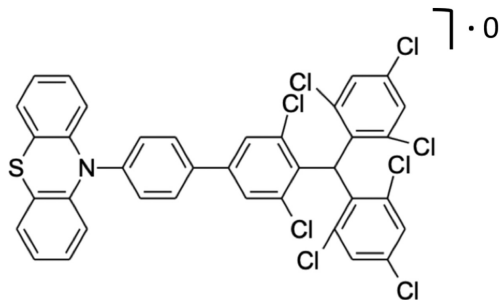


The unpaired molecular orbital associated with the radical character of the system lies energetically below one or several paired molecular orbitals.

SHI radical formation^[1]

Scenario 1: Spatially Disjoint Orbitals

Example: TTM-PPTA System



Initial closed-shell state:

- HOMO localised on TTM fragment
- HOMO-1 localised on PPTA fragment
- Minimal overlap leads to weak Coulomb repulsion between them

Upon ionisation:

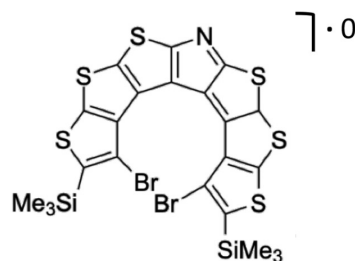
- electron removed from the HOMO
- significant reduction in Coulomb repulsion in the HOMO
- SOMO (formerly HOMO) energy drops significantly

Result:

- SOMO energy stabilises below the HOMO-1

Scenario 2: Nearly Degenerate Orbitals

Example: ATH System



Initial closed-shell state:

- HOMO and HOMO-1 quasi degenerate.
- Significant overlap between the HOMO and HOMO-1
- strong 'self-Coulomb' repulsion

Design strategies

- choosing appropriate molecular fragments to ensure spatial disjointness of frontier orbitals
- tuning of the SOMO–HOMO energy gap in a multifragment SHI radical *via* optimisation of the HOMO energies of the precursor molecules
- weak electrostatic repulsion between the HOMO and other occupied frontier MOs (small energy drop of other MO upon ionisation)
- strong repulsion between the α and β spin components in HOMO ('self-Coulomb' repulsion) of the closed-shell parent compound (large energy drop of other MO upon ionisation)
- computational prescreening and "big data" machine-learning approaches

HOMO and HOMO-1 of the CS spatially disjoint (minimal overlap) and quasi degenerate (effectively the same energy levels)

Applications

- organic luminescent emitters light-emitting diodes (OLEDs) devices
- increased photostability of SHI radicals versus non-SHI radicals^[2]

Methodology

- spin-(un)restricted KS DFT
- PBE0-D3/def2-SV(P)
- solvent corrections (PCM) for several solvents